SHORT COMMUNICATIONS

Contributions intended for publication under this heading should be expressly so marked; they should not exceed about 1000 words; they should be forwarded in the usual way to the appropriate Co-editor; they will be published as speedily as possible.

Acta Cryst. (1983). C39, 1473

N-Methylacetamide hemihydrochloride: corrigendum.* By RICHARD E. MARSH, Arthur Amos Noyes Laboratory of Chemical Physics, California Institute of Technology, Pasadena, California 91125, USA

(Received 16 May 1983; accepted 1 June 1983)

Abstract

The crystal structure of $C_3H_7NO.\frac{1}{2}HCl$ should be described in the orthorhombic space group *Fdd2* rather than the monoclinic *C2* reported by Jaber, Guilhem & Loiseleur [*Acta Cryst.* (1983), C**39**, 485–487].

The crystal structure derived for *N*-methylacetamide hemihydrochloride was based on the monoclinic space group C2 with a = 8.454 (4), b = 10.597 (5), c = 11.754 (5) Å, $\beta =$ 111.09 (8)°, Z = 8 (Jaber, Guilhem & Loiseleur, 1983; JGL). The vectors |102|, |100|, |010| define an effectively orthorhombic unit cell with a' = 21.933, b' (= a) = 8.454, c'(= b) = 10.597 Å, $a' = \beta' = 90$, $\gamma' = 90.01^\circ$, Z = 16. The corresponding transformations $x' = \frac{1}{2}z$, $y' = x - \frac{1}{2}z$, z' = ylead to atomic coordinates that are consistent with space group *Fdd2* within their reported uncertainties. The *Fdd2* coordinates are given in Table 1.

Since the increase in symmetry from C2 to Fdd2 does not involve the addition of a center of symmetry, the coordinate shifts necessary to achieve the higher symmetry are no larger

* Contribution No. 6843 from the Arthur Amos Noyes Laboratory of Chemical Physics. This work was supported by National Institutes of Health Research Grant No. GM 16966.

0108-2701/83/101473-01\$01.50

than the formal e.s.d.'s (Schomaker & Marsh, 1979) and the molecular dimensions as found in C2 (JGL) remain essentially unchanged; all molecular units become equivalent by symmetry.

Table 1. Atom coordinates $(\times 10^4)$ (space group Fdd2)

The C2 coordinates (JGL, Table 1) have been averaged according to the symmetry of Fdd2; numbers in square brackets are shifts necessary to achieve this symmetry, and numbers in parentheses are averaged e.s.d.'s in the original values.

	x'	y'	z'
Cl (1,2)	0	0	5012 01 (1)
C(1,4)	788 2 (1)	3730 [4] (3)	3143 31 (3)
N(1,2)	988 [0] (1)	2342 [0] (2)	3836 [2] (2)
C(2,5)	1560 1 (1)	1995 [4] (2)	4029 [0] (2)
O(1,2)	1970 1 (1)	2916 3 (2)	3614 2 (2)
C(3,6)	1714 0 (1)	533 [1] (3)	4745 [0] (2)

References

- JABER, M., GUILHEM, J. & LOISELEUR, H. (1983). Acta Cryst. C39, 485-487.
- SCHOMAKER, V. & MARSH, R. E. (1979). Acta Cryst. B35, 1933-1934.

© 1983 International Union of Crystallography

Acta Cryst. (1983). C 39, 1473-1474

Half-normal probability plot of 3,3'-dithiodipropionic acid. By M. SESHASAYEE, Department of Physics, Indian

Institute of Technology, Madras 600 036, India

(Received 26 May 1983; accepted 1 July 1983)

Abstract

A half-normal probability plot comparing the reported structural parameters of 3,3'-dithiodipropionic acid, $(SCH_2CH_2COOH)_2$, published by two different authors is presented.

0108-2701/83/101473-02\$01.50

Results of the crystal structure analysis of the title compound published by two different authors (Prout, 1982; Appa Rao, Seshasayee, Aravamudan, Nageswara Rao & Venkatasubramanian, 1982) have been compared by a half-normal probability plot (Abrahams & Keve, 1971) (Fig. 1):

© 1983 International Union of Crystallography

1474

 $\delta p_i = ||p(1)_i| - |p(2)_i|| / |\sigma^2 p(1)_i + \sigma^2 p(2)_i|^{\frac{1}{2}}$, where $p(1)_i$ and $p(2)_i$ are the two sets of parameters refined in the structure analysis and the σ 's are the corresponding standard deviations.

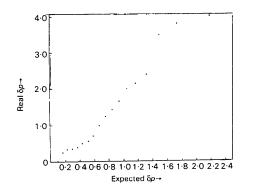


Fig. 1. Half-normal probability plot with real δp and expected δp as ordinate and abscissa.

A least-squares straight-line fit of the plot ignoring two points (16th and 17th in the figure) gave a slope of 1.83 (4) and an intercept of -0.16 (3). The non-zero intercept indicates systematic errors in the data. The slope of the straight line deviates from the ideal value of 1 by 0.85 indicating that either the $\Delta p_i = p(1)_i - p(2)_i$ are large or the corresponding standard deviations $\sigma(p_i)$ are small. The near linearity of the plot excludes the first possibility since for Δp_i to be large $p(1)_i$ and $p(2)_i$ must differ from the true value in opposite directions. The second possibility is more likely, which could arise if both sets of data are refined by least squares to a greater degree than warranted by the accuracy of the data.

References

ABRAHAMS, S. C. & KEVE, E. T. (1971). Acta Cryst. A27, 157-165.

APPA RAO, G. V. N., SESHASAYEE, M., ARAVAMUDAN, G., NAGESWARA RAO, T. & VENKATASUBRAMANIAN, P. N. (1982). Acta Cryst. B38, 2852–2855.

PROUT, K. (1982). Acta Cryst. B38, 338-340.

International Union of Crystallography

Chemical nomenclature

The attention of authors is drawn to the following recommendations of the IUPAC Commission on Nomenclature of Organic Chemistry: The Designation of Non-Standard Classical Valence Bonding in Organic Nomenclature [Pure Appl. Chem. (1982), **54**, 217–227], Revision of the Extended Hantzsch-Widman System of Nomenclature for Heteromonocycles [Pure Appl. Chem. (1983), **55**, 409–416]; and to the following recent recommendations of the IUPAC-IUB Joint Commission on Biochemical Nomenclature: Symbols for Specifying the Conformation of Polysaccharide Chains [Eur. J. Biochem. (1983), **131**, 5–7], Abbreviations and Symbols for the Description of Conformations of Polynucleotide Chains [Eur. J. Biochem. (1983), **131**, 9–15]. Authors are reminded that, wherever possible, chemical nomenclature in the Union's journals should conform to IUPAC rules. Basic rules for the nomenclature of inorganic chemistry are given in Nomenclature of Inorganic Chemistry (1970) ('The Red Book'), London: Butterworths and, for organic chemistry, in the two volumes Nomenclature of Organic Chemistry, Sections A,B,C,D,E,F, & H (1979) ('The Blue Book'), Oxford: Pergamon Press and Biochemical Nomenclature and Related Documents (1978), London: Biochemical Society. For additional details on nomenclature requirements see p. 179 of Notes for Authors [Acta Cryst. (1983), A**39**, 174–186].

An index to all IUPAC nomenclature publications is available from the IUPAC Secretariat, Bank Court Chambers, 2–3 Pound Way, Cowley Centre, Oxford OX4 3YF.